Unit 8 – Mixtures and Solutions

(Honors Chemistry Syllabus)

[Text Ref.: Ch. 3 (Sect. 3.3), Ch. 7 (Section 7.2 & 7.4), Ch. 8 (Section 8.5) & Ch. 14]

- 1. Mixture: (Section 14.1)
- **2. Heterogeneous Mixtures:** (Section 14.1)
- **3. 3 Types of Heterogeneous Mixtures:** (Section 14.1)
 - A. Suspensions
 - B. Thixotropic Mixtures
 - C. Colloids:
 - 1. Defn.
 - 2. Brownian motion
 - 3. Tyndall effect
- 4. Homogeneous Mixtures (Solutions):

(Sections 3.3, 7.2, 7.4, 14.1 & 14.4)

- A. Defn.
- B. 2 Parts of a Solution:
 - 1. Solvent
 - 2. Solute
- C. Soluble vs. Insoluble
- D. Types of Solutions:
 - 1. Gas solutions
 - 2. Liquid solutions
 - a. Aqueous solutions
 - b. Miscible vs. Immiscible
 - c. Electrolytes vs. Nonelectrolytes (*Hint:* Look on pgs. 215, 498 & 499)
 - d. Tincture (Hint: Not in book)
 - 3. Solid solutions
 - a. Alloy
 - b. Amalgam
- **5.** Chromatography (Section 3.3 & Lab 14-1)
- **6. Concentration,** []: (Section 14.2)
- **7. 5 ways to express** []: (Section 14.2)
 - A. Percent by mass
 - B. Percent by volume

- **7. 5 ways to express** []: (cont.)
 - C. Molarity(M) of a solution
 - D. Molality(m) of a solution
 - E. Mole Fraction(X) of a solution
- **8. The Solvation Process:** (Sections 8.5 & 14.3)
 - A. Solvation
 - B. Molecular Polarity
 - C. Solubility of polar and nonpolar molecules (Can you explain this phrase? "like dissolves like"?)
 - D. Solubility & hydration of ionic compounds in aqueous solutions

(<u>Hint:</u> Look on pgs. 268 & 490)

E. Solubility & hydration of molecules in aqueous solutions (*Hint:* Look on pgs. 268 & 491)

- **9. 3 Factors That Affect Solvation:** (Section 14.3)
 - A. Agitation
 - B. Surface Area
 - C. Temperature
- **10. Saturation of solutions:** (Section 14.3)
 - A. 3 Types of Saturated Solutions:
 - 1. Unsaturated solutions
 - 2. Saturated solutions
 - 3. Supersaturated solutions
 - B. Solubilities as a Function of Temperature
 - 1. Solubilities of Solids
 - 2. Solubilities of Gases
- 11. Concentration vs. Saturation (Not in book)
- **12. Colligative Properties:** (Section 14.4)
 - A. Defn.:
 - B. Boiling Point Elevation
 - C. Freezing Point Depression
 - D. Osmotic Pressure
- ✓ <u>Unit 8 Internet help (On Mr. Segulin's Honors Chemistry Web-page):</u>
 - Online Text
 - o CyberEd Chemistry: "Solutions" and "Solubility & Precipitation"
 - o General Chemistry help (pick the topic that you need to assist you)
- ✓ Ch. 14 Review Questions:
 - o Pgs. 474-511: #'s 1-4, 6-35, 39-43, 45-53, 54-88, 91, 95-111, 118, & 120.
 - o Pgs. 985-986: #'s 1-16, 19-21