Date: \_\_\_\_\_ AP Chemistry

# **Chapter 5 – Thermochemistry (Syllabus)**

**Chemistry: The Central Science** 

1. Read pgs. 163-211 noting the terms in bold and italics. Read and take notes on 5.1-5.2 before our next class.

### 2. <u>Intro. & 5.1 (The Nature of Energy):</u>

- Pay close attn. to the terms & concepts in this section. They set the foundation for the "chemical" applications later in the chapter.
- o Just read through the "physics" equations, but we will not focus on these calculations at this time (Those are for Physics class).
- o What are some units for energy?
- o The terms "system" & "surroundings" are important concepts to understand for later.

# 3. <u>5.2 (The First Law of Thermodynamics):</u>

- > One of the most important laws in science
- Pay attn. to the "+ & -" signs in <u>all</u> of the equations.
- > Endo- vs. Exothermic rxns.
- ➤ What are "state functions"?

## 4. <u>5.3 (Enthalpy) & 5.4 (Enthalpies of Reaction):</u>

- ✓ It's "pretty much" just heat flow.
- ✓ When is  $\Delta H$  positive? negative?
- ✓ Pay close attn. to the 3 guidelines when using thermochemical eqns. & enthalpy diagrams.

## 5. <u>5.5 (Calorimetry):</u>

- o We did "much" of this last year. The major equation should look familiar.
- o Again, watch your signs (& units)!!
- o What are the 2 types of calorimeters?

# 6. 5.6 (Hess's Law) & 5.7 (Enthalpies of Formation):

- \* "Really" understand this law!! We'll be using it in later chapters & it's an excellent AP test item.
- ➤ The math is easy (+ & -), but it's sometimes tough to figure out how to manipulate the equations in the problems.
- ➤ What is the standard enthalpy of formation of an element? Explain.
- Equation 5.31 is the most important equation in this section.

#### 7. 5.8 (Bond Enthalpies):

- Strength of bonds: Table 5.4
- $\Delta H_{rxn}$  can be estimated from bond energies
- 8. <u>5.9 (Foods and Fuels):</u> Read through. This includes "mostly" qualitative applications of enthalpy.

#### 9. <u>Labs:</u>

- ❖ Lab 5-5: Calorimetry and The Specific Heat of a Metal
- ❖ LabQuest 13: Heat of Neutralization and Hess's Law

#### 10. Chapter 5 Exercises:

- ✓ You are responsible for ALL of the "Visualizing Concepts" & "Exercises" at the end of Ch. 5.
- ✓ Remember to start with the Focus Problems (given below) and then progress from there, ending with the "Additional & Integrative Exercises".
- ✓ Ch. 5 Focus Problems:
  - ♦ <u>Visualizing Concepts & Exercises</u>: #'s 3, 4, 13, 17, 22, 25, 30, 37, 39, 43, 47, 51, 53, 55, 57, 65, 69, 71, 73, 77, 81, 83, 93, & 95
  - ♦ Additional Exercises: #'s 105, 106, 107, 109, & 115
  - ♦ <u>Integrative Exercises</u>: #'s 118, 119, 120, 123, & 125
- 11. <u>Test:</u> Chapter 5 test will be in about 2 weeks.